

## Talarolutins A–D: Meroterpenoids from an Endophytic Fungal Isolate of *Talaromyces minioluteus*

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### Abstract:

Four meroterpenoids [talarolutins A–D] and one known compound [purpurquinone A] were characterized from an endophytic fungal isolate of *Talaromyces minioluteus* (G413), which was obtained from the leaves of the medicinal plant milk thistle [*Silybum marianum* (L.) Gaertn. (Asteraceae)]. The structures of talarolutins A–D were determined by the analysis of various NMR and MS techniques. The relative and absolute configuration of talarolutin A was determined by X-ray diffraction analysis. A combination of NOESY data and comparisons of ECD spectra were employed to assign the relative and absolute configuration of the other analogs. Talarolutins B–D were tested for cytotoxicity against human prostate carcinoma (PC-3) cell line, antimicrobial activity, and induction of quinone reductase; no notable bioactivity was observed in any assay.

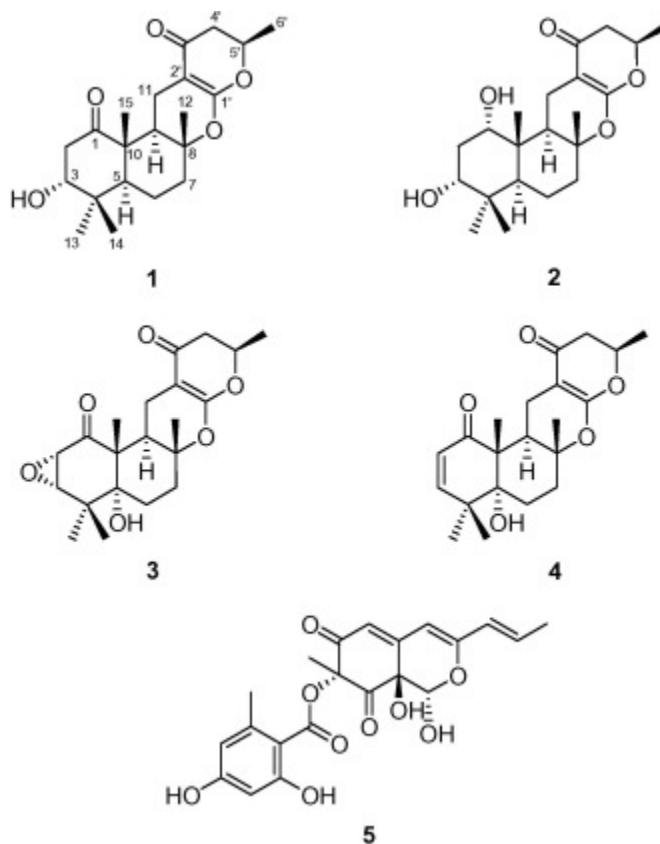
**Keywords:** Meroterpenoid | Fungal endophyte | *Talaromyces minioluteus* | Milk thistle | *Silybum marianum* | Asteraceae

### Article:

#### 1. Introduction

Plant-based secondary metabolites (primarily flavonolignans) from the medicinal herb milk thistle [*Silybum marianum* (L.) Gaertn. (Asteraceae)] have been studied extensively (Althagafy et al., 2013, Davis-Searles et al., 2005, Graf et al., 2007, Gufford et al., 2014, Napolitano et al., 2013, Polyak et al., 2010, Sy-Cordero et al., 2013). Recent studies that have explored the chemical and fungal diversity of endophytes associated with this plant have also yielded interesting results (El-Elimat et al., 2014b, Figueroa et al., 2014, Raja et al., 2015). In continuation of these investigations in search of new and/or biologically active natural products,

an endophytic fungal isolate of *Talaromyces minioluteus* (G413), which was isolated from the leaves of milk thistle, was subjected to natural products chemistry techniques to yield four new meroterpenoids (Fig. 1), named here as talarolutins A–D (**1–4**), and one known compound, purpurquinone A (**5**) (Wang et al., 2011). Structurally diverse meroterpenoids with antimicrobial, antiviral, antitumor, immunomodulatory, and phytotoxic effects have been reported previously from various fungal sources (Geris and Simpson, 2009). Thus, compounds **2–4** were examined for biological activity in three assays, including cytotoxicity against a human prostate carcinoma (PC-3) cell line, antimicrobial activity, and induction of quinone reductase, but were all inactive.



**Fig. 1.** Structures of talarolutins A–D (**1–4**) and purpurquinone A (**5**).

## 2. Results and discussion

Talarolutin A (**1**) had the molecular formula  $C_{21}H_{30}O_5$ , yielding an index of hydrogen deficiency of seven. Signals for five methyl groups (four singlets and one doublet), two oxygenated methines, and a series of aliphatic protons, including five methylene and two methine units, were observed in the  $^1H$  NMR spectrum of **1** (Fig. S1, Supporting information and Table 1). In addition to the signals expected for these structural features, the  $^{13}C$  NMR spectrum showed resonances for three quaternary and four non-protonated  $sp^2$ -hybridized carbons (Fig. S2 and Table 2). The sesquiterpenoid-derived portion of **1** was constructed by the analysis of COSY, HSQC, and HMBC NMR data (Table 1, Table 2, and S1). The upfield shifted methyl singlets, H<sub>3</sub>-14 ( $\delta_H$  1.08, 3H) and H<sub>3</sub>-13 ( $\delta_H$  1.02, 3H), showed common HMBC correlations to an oxygenated carbon, C-3 ( $\delta_C$  80.0), a quaternary carbon, C-4 ( $\delta_C$  38.4), and C-5 ( $\delta_C$  51.4). Further HMBC correlations from H-3 ( $\delta_H$  3.88, br s) to a ketone carbonyl, C-1 ( $\delta_C$  211.5) and C-

5, as well as, COSY correlations to H<sub>2</sub>-2 ( $\delta_{\text{H}}$  3.35, dd,  $J$  = 13.5, 3.2 Hz and 2.22, dd,  $J$  = 13.5, 3.7 Hz, 2H) were observed. These data, in conjunction with HMBC correlations from methyl singlet H<sub>3</sub>-15 ( $\delta_{\text{H}}$  1.24, 3H) to C-1, C-5, C-9 ( $\delta_{\text{C}}$  44.6), and quaternary carbon, C-10 ( $\delta_{\text{C}}$  51.7), supported the presence of a cyclohexanone ring system. COSY NMR data identified a spin system H-5/H<sub>2</sub>-6/H<sub>2</sub>-7 (Fig. S3), including two methylene units ( $\delta_{\text{H}}$  1.69 and 1.50, 2H, for H<sub>2</sub>-6;  $\delta_{\text{H}}$  2.04 and 1.66, 2H, for H<sub>2</sub>-7). HMBC correlations from methyl protons, H<sub>3</sub>-12 ( $\delta_{\text{H}}$  1.26, s, 3H) to C-7 ( $\delta_{\text{C}}$  39.7), an oxygenated quaternary carbon, C-8 ( $\delta_{\text{C}}$  83.4), and C-9 completed the decalin portion of the ring system of **1**.

**Table 1.** <sup>1</sup>H NMR spectroscopic data (400 MHz) for **1–4** in CDCl<sub>3</sub>.

#	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>
	$\delta_{\text{H}}$ (mult., $J$ )	$\delta_{\text{H}}$ (mult., $J$ )	$\delta_{\text{H}}$ (mult., $J$ )	$\delta_{\text{H}}$ (mult., $J$ )
1		3.70 (br s)		
2 $\alpha$	3.35 (dd, 13.5, 3.2)	2.19 (dt, 13.5, 2.5)	3.48 (d, 3.3)	5.76 (d, 10.3)
2 $\beta$	2.22 (dd, 13.5, 3.7)	1.96 (dt, 13.5, 2.7)		
3	3.88 (br s)	3.52 (br s)	3.27 (d, 3.3)	6.17 (d, 10.3)
5	1.81 (dd, 12.7, 2.0)	1.78 (dd, 12.6, 2.0)		
6 $\alpha$	1.69 (m)	1.74 (m)	1.75 (dt, 13.9, 3.5)	1.79 (dd, 14.4, 3.6)
6 $\beta$	1.50 (m)	1.41 (m)	1.60 (m)	1.73 (ddd, 14.4, 4.1, 3.3)
7 $\alpha$	2.04 (m)	2.06 (m)	2.21 (m)	2.19 (dt, 4.9, 14.1)
7 $\beta$	1.66 (m)	1.70 (m)	1.70 (dt, 12.8, 3.4)	1.82 (m)
9	2.04 (m)	2.21 (dd, 13.0, 4.9)	2.64 (dd, 12.8, 4.5)	2.60 (dd, 12.8, 4.5)
11 $\alpha$	2.96 (dd, 15.5, 4.5)	2.37 (dd, 15.2, 4.9)	3.04 (dd, 15.3, 4.5)	3.15 (dd, 15.5, 4.5)
11 $\beta$	1.91 (dd, 15.5, 12.7)	1.99 (dd, 15.2, 13.0)	1.98 (dd, 15.3, 12.8)	2.01 (dd, 15.5, 12.8)
12	1.26 (s)	1.25 (s)	1.25 (s)	1.33 (s)
13	1.02 (s)	1.01 (s)	1.29 (s)	1.25 (s)
14	1.08 (s)	0.83 (s)	1.27 (s)	1.13 (s)
15	1.24 (s)	0.83 (s)	1.23 (s)	1.297 (s)
4' $\alpha$	2.48 (dd, 16.9, 14.0)	2.49 (dd, 16.8, 14.0)	2.48 (dd, 16.9, 14.4)	2.49 (dd, 16.9, 14.0)
4' $\beta$	2.35 (dd, 16.9, 3.2)	2.32 (dd, 16.8, 3.1)	2.36 (dd, 16.9, 2.9)	2.36 (dd, 16.9, 3.2)
5'	4.57 (dq, 14.0, 6.3, 3.2)	4.54 (ddq, 14.0, 3.1, 6.3)	4.60 (ddq, 14.4, 2.9, 6.3)	4.60 (ddq, 14.0, 3.2, 6.3)
6'	1.43 (d, 6.3)	1.42 (d, 6.3)	1.43 (d, 6.3)	1.44 (d, 6.3)
3-OH	1.54 (br s)			
5-OH			2.68 (br d, 2.0)	1.44 (s)

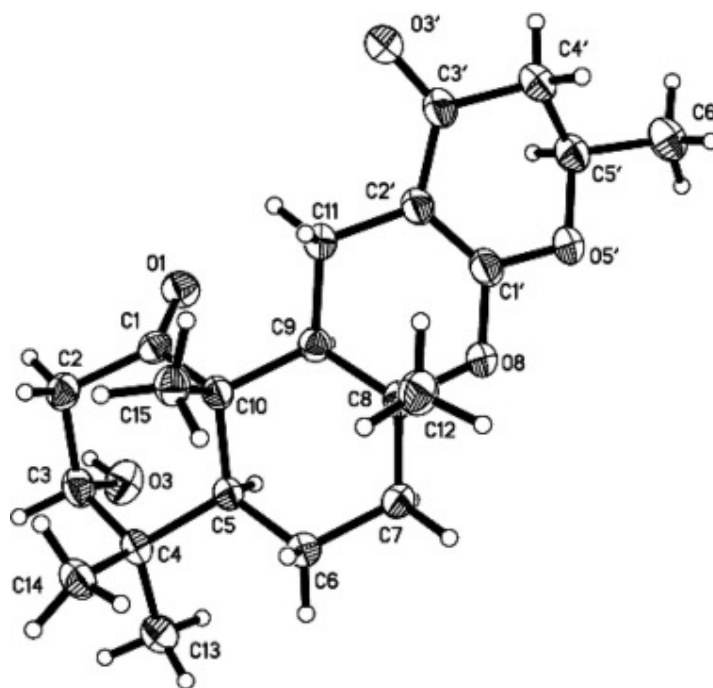
To complete the structural assignments of **1**, HMBC correlations from H<sub>2</sub>-11 ( $\delta$  2.96, dd,  $J$  = 15.5, 4.5 Hz and 1.91, dd,  $J$  = 15.5, 12.7 Hz, 2H) to C-8, C-9, C-10, olefinic carbons, C-1' ( $\delta_{\text{C}}$  168.4) and C-2' ( $\delta_{\text{C}}$  91.1), and ketone carbonyl carbon, C-3' ( $\delta_{\text{C}}$  191.2) appended the  $\alpha,\beta$ -unsaturated carbonyl unit to the bicyclic ring, thereby securing the attachment of C-11 to C-2'. The placement of a methylene unit, H<sub>2</sub>-4' ( $\delta_{\text{H}}$  2.48, dd,  $J$  = 16.9, 14.0 Hz and  $\delta$  2.35, dd,  $J$  = 16.9, 3.2 Hz, 2H), alpha to the ketone (C-3') was supported by HMBC correlations from H<sub>2</sub>-4' to C-2' and C-3'. Multiplicity of H<sub>2</sub>-4', COSY NMR data, and additional HMBC correlations from H<sub>2</sub>-4' to C-5' ( $\delta_{\text{C}}$  75.8) and C-6' ( $\delta_{\text{C}}$  20.7) enabled the identification of the spin system, H<sub>2</sub>-4'/H-5'/H<sub>3</sub>-6', including the methyl group ( $\delta_{\text{H}}$  1.43, d,  $J$  = 6.3 Hz, 3H, for H<sub>3</sub>-6'). Although an HMBC correlation from H-5' ( $\delta_{\text{H}}$  4.57, dq,  $J$  = 14.0, 6.3, 3.2 Hz) to C-1' was not observed, the chemical shifts of C-1' and C-5' were consistent with an ether linkage between the two carbons, completing the modified pyranone-type ring system. A second ether linkage between carbons C-8 and C-1' accounted for the last remaining unsaturation, as well as the significant downfield chemical shift of C-1' ( $\delta_{\text{C}}$  168.4). Attempts to assign the absolute configuration of **1** using

Mosher's method resulted in degradation of the compound (Hoye et al., 2007). Fortunately, X-ray diffraction analysis of a crystal obtained during the course of these studies not only confirmed the structure (Fig. 2), but also enabled the unambiguous assignment of the absolute configuration of **1** by employing the Flack parameter [ $F = 0.01$  (18)] (Parsons and Flack, 2004).

**Table 2.**  $^{13}\text{C}$  NMR spectroscopic data (100 MHz) for **1–4** in  $\text{CDCl}_3$ .<sup>a</sup>

#	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>
1	211.5	72.1	202.7	203.0
2	42.8	29.8	54.5	124.2
3	80.0	77.8	64.0	150.7
4	38.4	38.0	37.7	42.0
5	51.4	42.3	76.7	78.4
6	19.6	19.4	25.1	25.2
7	39.7	40.0	31.9	32.8
8	83.4	84.7	82.8	82.9
9	44.6	44.5	38.7	38.7
10	51.7	41.5	56.9	53.6
11	17.1	15.3	17.8	17.7
12	20.9	20.71	20.2	20.5
13	27.7	28.5	23.8	24.0
14	22.2	21.8	23.4	25.6
15	14.5	15.6	16.2	18.6
1'	168.4	168.9	168.4	168.4
2'	91.1	90.5	91.1	91.5
3'	191.2	191.5	191.3	191.3
4'	42.9	42.7	42.8	42.8
5'	75.8	75.6	75.9	75.8
6'	20.7	20.67	20.6	20.6

<sup>a</sup>  $\delta$  in ppm.



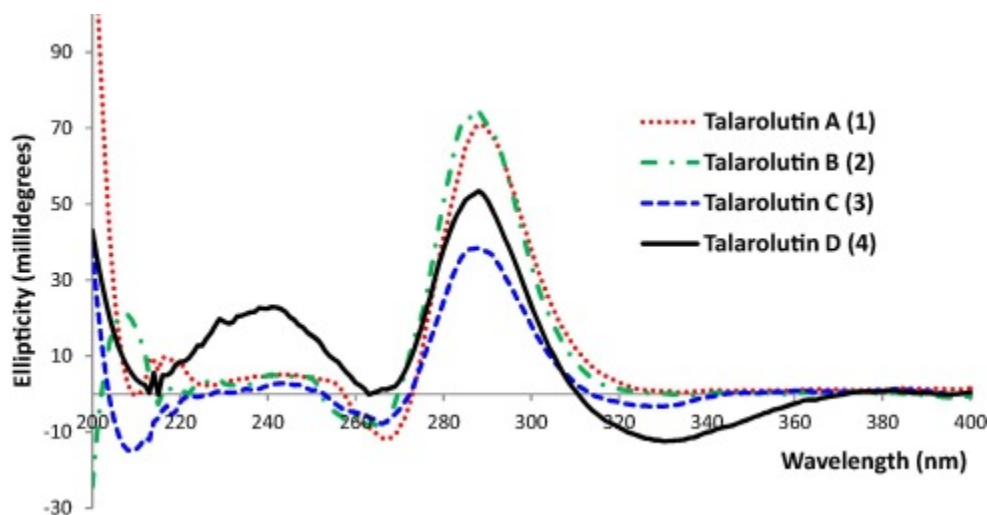
**Fig. 2.** ORTEP drawing of the molecular structure of **1**.

The molecular formula of talarolutin B (**2**) was determined to be  $C_{21}H_{32}O_5$  (index of hydrogen deficiency of six) on the basis of HRESIMS data. The  $^1H$  and  $^{13}C$  NMR spectra of **2** (Figs. S4, S5 and Table 1, Table 2) closely resembled those of **1**. However, an additional oxygenated methine signal ( $\delta_C/\delta_H$  72.1/3.70 for H-1) was observed in the NMR spectra of **2** that replaced the ketone carbonyl carbon observed in the  $^{13}C$  NMR spectrum of **1** ( $\delta_C$  211.5 for C-1). These changes were consistent with the reduction of the carbonyl group at position C-1 to a secondary alcohol in **2** and accounted for the only key difference in the structure of talarolutin B (**2**), including one less degree of unsaturation. The small  $J$ -values observed for H-1, resulting in a broad peak in the  $^1H$  NMR spectrum (Fig. S4) supported the axial orientation of the hydroxy group. NOESY correlations between H-1 and H<sub>3</sub>-15/H<sub>3</sub>-14 ( $\delta_H$  0.83; overlapping signals) as well as between H-3, H<sub>3</sub>-14 ( $\delta_H$  0.83), and H<sub>3</sub>-13 ( $\delta_H$  1.01) were also consistent with a 1, 3-diaxial orientation for the hydroxy groups in **2**. Assuming identical relative and absolute configuration at the in-common asymmetric centers between **1** and **2** (based on biosynthetic origins), C-1 was assigned the *S*-configuration by analysis of the proton coupling constants and NOESY NMR data (Fig. S6).

Inspection of the  $^1H$  NMR spectrum (Fig. S7 and Table 1) of talarolutin C (**3**) showed signals corresponding to the structural features similar to those observed for **1** and **2**. The presence of a pair of coupled doublets ( $\delta_H$  3.48, d,  $J$  = 3.3 Hz for H-2 and  $\delta_H$  3.27, d,  $J$  = 3.3 Hz for H-3), characteristic of epoxide protons in the  $^1H$  NMR spectrum of **3**, accounted for one of the key differences. Corresponding carbon signals ( $\delta_C$  54.5 for C-2 and  $\delta_C$  64.0 for C-3) were also observed in the  $^{13}C$  NMR spectrum (Fig. S8 and Table 2). The position of the C-2/C-3 epoxide group was supported by HMBC correlations from H-2 to C-1 ( $\delta_C$  202.7) and C-10 ( $\delta_C$  56.9), as well as from H-3 to C-1, C-4 ( $\delta_C$  37.7), C-5 ( $\delta_C$  76.7), and C-14 ( $\delta_C$  23.4). A NOESY correlation between H-2 and H-3 suggested a *syn*-epoxide moiety, and an additional correlation between H-2 and H<sub>3</sub>-15 ( $\delta_H$  1.23) placed these protons on the same face of the six-membered ring system. Additionally, another difference between **1** and **3** was oxygenation at C-5 in **3** relative to **1**, resulting in a downfield-shift of C-5 ( $\delta_C$  76.7) in **3** relative to **1** ( $\delta_C$  51.4) and the appearance of an additional hydroxy group ( $\delta_H$  2.68 for 5-OH) at the ring junction; the HRESIMS data [ $m/z$  377.1942 ( $M+H$ )<sup>+</sup>] were consistent with the molecular formula ( $C_{21}H_{28}O_6$ ) of talarolutin C (**3**), including an extra oxygen moiety relative to the formula of **1**. NOESY correlations between H-6 $\alpha$  ( $\delta_H$  1.75), H<sub>3</sub>-13 ( $\delta_H$  1.29), and 5-OH, as well as from methylene proton, H-6 $\beta$  ( $\delta_H$  1.60) to pseudo-axially oriented methyl groups, H<sub>3</sub>-15, H<sub>3</sub>-14 ( $\delta_H$  1.27), and H<sub>3</sub>-12 ( $\delta_H$  1.25) supported the *trans*-ring junction for terpene-derived bicyclic ring system (Fig. S6); while not definitive, the absence of NOESY correlations between 5-OH and H<sub>3</sub>-15 were consistent with this assignment. The remaining asymmetric centers that were common with **1** and **2** were assigned analogous absolute configurations, as discussed previously.

In the  $^1H$  and  $^{13}C$  NMR spectra (Figs. S9, S10 and Table 1, Table 2) of talarolutin D (**4**), the epoxide signals observed in **3** were replaced by resonances characteristic of a double bond ( $\delta_C/\delta_H$  124.2/5.76, d,  $J$  = 10.3 and  $\delta_C/\delta_H$  150.7/6.17, d,  $J$  = 10.3 for C-2/H-2 and C-3/H-3, respectively). The molecular formula of **4** was determined to be  $C_{21}H_{28}O_5$  (eight unsaturations), which was consistent with such a substitution. The remaining structural features were similar to **3** and since no new asymmetric centers were generated, an analogous relative configuration was proposed for **4**. A SciFinder search showed compounds **3** and **4** to be a part of an industrial

catalog. However, lack of reported NMR data as well as absence of relative and absolute configuration assignment did not allow unambiguous identification of these compounds. Therefore, **3** and **4** are considered new natural products and have been fully characterized in this report. The experimental ECD spectra (Fig. 3) for **1–4** showed a comparable trend supporting the assignment of all talarolutin analogs to the same enantiomeric series.



**Fig. 3.** Experimental ECD spectra of **1–4**.

As with previous research on compounds isolated from endophytes of milk thistle (Raja et al., 2015), the compounds **2–4** were evaluated for cytotoxicity against the human prostate carcinoma (PC-3) cell line but all were inactive ( $IC_{50}$  values  $>25 \mu M$ ). They were also examined in a suite of antimicrobial assays, but did not show any effects (MIC values  $>100 \mu g/mL$ ) against *Staphylococcus aureus*, *Escherichia coli*, *Mycobacterium smegmatis*, *Candida albicans*, and *Aspergillus niger*. The structural features of talarolutins bore close resemblance to recently reported marine natural products, penicillipyrones A and B (Liao et al., 2014). The latter of these was reported to show significant induction of quinone reductase. Therefore, compounds **2–4** were also tested for biological activity in this assay but were found to be ineffective at the highest concentration tested ( $25 \mu M$ ). Due to low sample amounts, compounds **1** and **5** were not evaluated for any biological activity in various assays. Compound **5** was originally isolated from *Talaromyces purpureogenus* and tested for bioactivity against influenza A virus (H1N1) but was found to be inactive (Wang et al., 2011).

### 3. Conclusion

Examination of an endophytic fungal isolate of milk thistle resulted in isolation of four new natural products from *T. minioluteus*. Biosynthetically, these compounds appear to be assembled by a combination of terpene-derived farnesyl moiety and a pyrone unit (Liao et al., 2014), and there is a growing interest in the biosynthesis of meroterpenoids due to their hybrid terpene and polyketide origins (Itoh et al., 2010). Although inactive in a few available assays, the new compounds append to only a handful of other reports of the chemistry from *T. minioluteus* (Ngokpol et al., 2015, Yilmaz et al., 2014). The mycology of the genus *Talaromyces*, in addition to a brief summary of the few secondary metabolites reported from *T. minioluteus*, has been reviewed recently (Yilmaz et al., 2014). Miniolins A–C were reported from an epigenetically

modified culture of *Penicillium minioluteum* (now *T. minioluteus*) (Tang et al., 2015). Our ongoing studies of endophytes of milk thistle and other medicinal plants have also yielded a suite of secondary metabolites belonging to structurally diverse classes (Bussey et al., 2015, El-Elmat et al., 2014b, Figueroa et al., 2014, Kaur et al., 2016, Raja et al., 2015). These observations support the overall goal of exploring this ecological group for new secondary metabolites.

## 4. Experimental

### 4.1. General experimental procedures

NMR experiments were conducted using an Agilent-700 and/or JEOL ECS-400 spectrometers (700 or 400 MHz for  $^1\text{H}$  NMR and 175 or 100 MHz for  $^{13}\text{C}$  NMR; Agilent Technologies, Santa Clara, CA, USA; JEOL Ltd., Tokyo, Japan). HRESIMS data were collected using an electrospray ionization (ESI) source coupled to a Q-ToF Premier mass spectrometer (Waters Corp., Milford, MA, USA) or a LTQ Orbitrap XL system (Thermo Fisher Scientific, San Jose, CA, USA) in positive and/or negative ionization modes by direct injection or *via* a liquid chromatography/autosampler system comprised of Acquity UPLC system (Waters Corp.). A CombiFlash Rf system using a RediSep Rf Si-gel Gold column (both from Teledyne-Isco, Lincoln, NE, USA) was employed for normal phase flash column chromatography (CC). HPLC separations were performed utilizing a Varian Prostar HPLC system (Varian Inc., Palo Alto, CA, USA) equipped with Prostar 210 pumps and a Prostar 335 photodiode array detector, using Galaxie Chromatography Workstation software (version 1.9.3.2, Varian Inc.). Kinetex  $\text{C}_{18}$  (Phenomenex, Torrance, CA, USA; 5  $\mu\text{m}$ ; columns of dimensions  $250 \times 21.2$  mm and  $250 \times 4.6$  mm) were used for preparative and analytical HPLC. YMC ODS-A (Waters Corp.; 5  $\mu\text{m}$ ; columns of dimensions  $250 \times 10$  mm and  $250 \times 4.6$  mm) columns were used for semi-preparative and analytical HPLC in selected cases. For UPLC analysis, a BEH  $\text{C}_{18}$  (Waters Corp.; 1.7  $\mu\text{m}$ ;  $50 \times 2.1$  mm) column was used. Optical rotation data were acquired on a Rudolph Research Autopol III polarimeter (Rudolph Research Analytical, Flanders, NJ, USA). ECD data were collected using an Olis DSM 17 CD spectrophotometer (Olis, Bogard, GA, USA). UV data were obtained using a Varian Cary 100 Bio UV-vis spectrophotometer (Varian Medical Systems, Palo Alto, CA, USA). IR data were collected using PerkinElmer Spectrum One with Universal ATR attachment (PerkinElmer, Inc., Waltham, MA, USA). The solvents were obtained from Fisher Scientific.

### 4.2. Isolation and identification of fungal strain

Fungal strain G413 was isolated as an endophyte from healthy surface sterilized leaves of milk thistle using procedures outlined previously (Figueroa et al., 2014). Based on morphology, it was evident that strain G413 belonged to *T. minioluteus* (Dierckx) Samson, Yilmaz, Frisvad, and Seifert (Yilmaz et al., 2014) (Fig. S11). Micromorphology of conidiophore and conidia agrees well with the original protologue presented recently in a polyphasic taxonomic study of the genus *Talaromyces* (Yilmaz et al., 2014).

For molecular identification of strain G413, two gene regions were sequenced. First, the nuclear ribosomal internal transcribed spacer region (ITS) was sequenced as it has been identified as a barcoding marker for fungi (Schoch et al., 2012). In addition to the ITS region, sequence data

was obtained from the RNA polymerase II largest subunit gene (*RPB1*). The *RPB1* region has been utilized in phylogenetic studies of subgenus *Biverticillium* (Samson et al., 2011). DNA extraction, PCR amplification, and phylogenetic analysis were performed following methods summarized earlier (El-Elimat et al., 2013, Figueroa et al., 2014). The *RPB1* region was amplified using primers RPB1-F1843 and RPB1-R3096 and PCR protocols outlined by Houbraken and Samson (2011) with some modifications. The first PCR reaction was carried out in 25 µL containing 3 µL template DNA, 2.5 µL BSA (New England BioLabs Inc), 2.5 µL 50% DMSO (Sigma), and 1 µL of each 10 µM forward (RPB1-F1843) and reverse (RPB1-R3096) primer. The rest of the volume was made up to 25 µL by adding molecular biology grade H<sub>2</sub>O from Fisher Scientific. After the first PCR, 1.5 µL PCR product was used to run a second PCR using the same protocol as the first PCR reaction. The PCR products were then run on an ethidium bromide-stained 1% agarose gel (Fisher Scientific) along with a 1 kb DNA ladder (Promega) to estimate the size of the amplified band. PCR products were finally purified using a Wizard SV Gel and PCR Clean-up System. Bidirectional Sanger sequencing of the purified PCR products was performed at Eurofins Genomics (<http://www.operon.com/default.aspx>) using BigDye Terminator v3.1 cycle sequencing. Sequences are edited using Sequencher 5.2.3 (Gene Codes Corp.). BLAST searches were then performed against the NCBI GenBank database separately using both the ITS region as well as the *RPB1* gene. Based on the ITS region BLAST search, against the TYPE database (Schoch et al., 2014), the closest hits were *T. minioluteus* (CBS 642.68; from TYPE material) GenBank NR\_121527.1; Identities = 561/575 (98%); Gaps = 12/572 (2%), *Penicillium samsonii* (CBS 137.84) GenBank JN899369; Identities = 562/571 (98%); Gaps = 9/571 (1%), and *Penicillium purogenum* var. *rubrisclerotium* (CBS 270.35) GenBank JN899381; Identities = 560/571 (98%); Gaps = 10/571 (1%). Similar results were obtained using the *RPB1* region. According to Samson et al., 2011, Yilmaz et al., 2014 *P. samsonii* and *P. purogenum* var. *rubrisclerotium* are synonymous with *T. minioluteus*. The *RPB1* sequences from the top BLAST search based on sequence similarity were downloaded and incorporated into an alignment with *RPB1* sequence from G413 for a maximum likelihood (ML) analysis using RAXML (Stamatakis, 2006). Results from both the BLAST search and ML analysis suggests that strain G413 can be identified as *T. minioluteus* [Pezizomycotina; Trichocomaceae]; the sequence from strain G413 forms a strong clade with 99% bootstrap statistical support with *T. minioluteus*, *P. purogenum* var. *rubrisclerotium*, and *P. samsonii* (Fig. S12). The fungal culture is maintained at the University of North Carolina at Greensboro, Department of Chemistry and Biochemistry Fungal Culture Collection, and a voucher specimen (G413) is deposited there. The sequence data were deposited in the GenBank (ITS: KM215653; *RPB1*: KU363961).

#### 4.3. Fermentation and extraction

For extractions, fungal cultures were grown on rice using procedures detailed previously (El-Elimat et al., 2014a). Briefly, seed cultures grown on malt extract agar (MEA; Difco) medium were excised from the leading edge of the colony and transferred to a liquid medium containing 2% soy peptone, 2% dextrose and 1% yeast extract (YESD). Following incubation (7 d) at 22 °C with agitation, the culture was used to inoculate rice media (50 mL) prepared using rice and twice the volume of rice with H<sub>2</sub>O in a 250 mL Erlenmeyer flask. This was incubated at 22 °C until the cultures showed good growth (14–21 d). To produce larger amounts of material, three



250 mL Erlenmeyer flasks were inoculated in an identical manner using one seed culture for each flask.

To each solid-substrate fermentation culture (G413) grown on rice (three flasks), CH<sub>3</sub>OH:CHCl<sub>3</sub> (60 mL, 1:1 v/v) was added. The culture was chopped into small pieces with a spatula and shaken overnight (~125 rpm at rt) using a rotary shaker. The sample was vacuum filtered, and the remaining residues were washed with small volumes of CH<sub>3</sub>OH:CHCl<sub>3</sub> (1:1 v/v). The filtrates from the three flasks were combined and CHCl<sub>3</sub>:H<sub>2</sub>O (540 mL, 1:1 v/v) were added, followed by stirring for 30 min. The organic layer was collected and evaporated to dryness under reduced pressure. The organic extract was further partitioned between CH<sub>3</sub>OH:CH<sub>3</sub>CN (300 mL, 1:1 v/v) and hexanes (300 mL). The CH<sub>3</sub>OH:CH<sub>3</sub>CN layer was evaporated to dryness in vacuum to yield crude extract (802 mg).

#### 4.4. Isolation

The organic extract (802 mg) was dissolved in a minimum amount of CHCl<sub>3</sub>:CH<sub>3</sub>OH (1:1 v/v), adsorbed onto Celite 545, and subjected to flash normal-phase CC using a gradient solvent system of hexane:CHCl<sub>3</sub>:CH<sub>3</sub>OH at 30 mL/min flow rate and 61 column volumes over 34.1 min to afford three fractions. Fraction 2 (70 mg) was subjected to preparative RP HPLC [Kinetex-C<sub>18</sub> column, linear gradient elution using CH<sub>3</sub>CN:H<sub>2</sub>O (containing 0.1% HCOOH): 40–80% CH<sub>3</sub>CN for 20 min;  $\lambda$  = 210 and 254 nm; 21.2 mL/min] to yield **4** (16.8 mg;  $t_R$  = 5.5 min) and a mixture of **3** and **4** (14.6 mg;  $t_R$  = 6.0 min). The mixture of **3** and **4** was then purified using prep RP HPLC [Kinetex-C<sub>18</sub> column, isocratic CH<sub>3</sub>CN:H<sub>2</sub>O (containing 0.1% HCOOH), 35:65 v/v, 21.2 mL/min] affording pure talarolutin C (**3**, 8.5 mg,  $t_R$  = 11.0 min) and talarolutin D (**4**; 3.0 mg;  $t_R$  = 9.5 min). Preparative RP HPLC [Kinetex-C<sub>18</sub> column, gradient elution using CH<sub>3</sub>CN:H<sub>2</sub>O (containing 0.1% HCOOH): 40–60% CH<sub>3</sub>CN for 20 min; 60–80% CH<sub>3</sub>CN for 10 min;  $\lambda$  = 210 and 254 nm; 21.2 mL/min] of fraction 3 (480 mg) resulted in isolation of **4** (14.7 mg;  $t_R$  = 6.0 min) and mixtures of **1** (12.0 mg;  $t_R$  = 6.5 min), **5** (11.7 mg;  $t_R$  = 7.0 min), and **2** (19.5 mg;  $t_R$  = 9.5 min). Compounds **1**, **2**, and **5** were further purified using prep RP HPLC [Kinetex-C<sub>18</sub> column, 21.2 mL/min] affording pure talarolutin A [**1**, 1.7 mg,  $t_R$  = 11.5 min; isocratic CH<sub>3</sub>CN:H<sub>2</sub>O (containing 0.1% HCOOH), 35:65 v/v], talarolutin B [**2**, 6.9 mg,  $t_R$  = 12.5 min; isocratic CH<sub>3</sub>CN:H<sub>2</sub>O (containing 0.1% HCOOH) 35:65 v/v], and purpurquinone A [**5**, 1.2 mg,  $t_R$  = 12.5 min; isocratic CH<sub>3</sub>CN:H<sub>2</sub>O (containing 0.1% HCOOH) 40:60 v/v]. The HRMS and <sup>1</sup>H NMR data for **5** were fully consistent with those reported in literature (Wang et al., 2011). To achieve ≥95% pure sample of **2**, an additional purification step employing semi-preparative RP HPLC [YMC-C<sub>18</sub> column, isocratic CH<sub>3</sub>CN:H<sub>2</sub>O (containing 0.1% HCOOH), 40:60 v/v;  $\lambda$  = 210 and 254 nm; 3 mL/min] was carried out to yield pure talarolutin B (**2**, 5.7 mg,  $t_R$  = 29.0 min).

#### 4.5. Talarolutin A (**1**)

Colorless crystals from CH<sub>3</sub>OH:CH<sub>3</sub>CN (1:1 v/v); [ $\alpha$ ]<sub>D</sub><sup>23</sup> = +237 (c = 0.05, CH<sub>3</sub>OH); UV/Vis (CH<sub>3</sub>OH)  $\lambda_{max}$  (log  $\epsilon$ ) 205 (3.1), 272 (4.0) nm; ECD (92  $\mu$ M, CH<sub>3</sub>OH)  $\lambda_{max}$  ( $\Delta\epsilon$ ) 267 (–12), 289 (+71) nm; for <sup>1</sup>H, <sup>13</sup>C, and HMBC NMR spectroscopic data, see Table 1, Table 2, and S1 respectively; HRESIMS  $m/z$  363.2151 [M+H]<sup>+</sup> (calcd. for C<sub>21</sub>H<sub>31</sub>O<sub>5</sub>, 363.2166).

#### 4.6. Talarolutin B (**2**)

Colorless powder;  $[\alpha]_{\text{D}23} = +200$  ( $c = 0.11$ , CH<sub>3</sub>OH); UV/Vis (CH<sub>3</sub>OH)  $\lambda_{\text{max}}$  (log  $\epsilon$ ) 208 (3.0), 271 (3.7) nm; IR (diamond)  $\nu_{\text{max}}$  2971, 1581, 1423, 1052, 1033 cm<sup>-1</sup>; ECD (183  $\mu\text{M}$ , CH<sub>3</sub>OH)  $\lambda_{\text{max}}$  ( $\Delta\epsilon$ ) 266 (−9), 287 (+75) nm; for <sup>1</sup>H, <sup>13</sup>C, and HMBC NMR spectroscopic data, see Table 1, Table 2, and S1 respectively; HRESIMS  $m/z$  365.2306 [M+H]<sup>+</sup> (calcd. for C<sub>21</sub>H<sub>33</sub>O<sub>5</sub>, 365.2322).

#### 4.7. Talarolutin C (3)

Colorless powder;  $[\alpha]_{\text{D}23} = +124$  ( $c = 0.15$ , CH<sub>3</sub>OH); UV/Vis (CH<sub>3</sub>OH)  $\lambda_{\text{max}}$  (log  $\epsilon$ ) 211 (2.6), 271 (3.6) nm; IR (diamond)  $\nu_{\text{max}}$  2998, 1587, 1422, 1186, 1139, 1052 cm<sup>-1</sup>; ECD (177  $\mu\text{M}$ , CH<sub>3</sub>OH)  $\lambda_{\text{max}}$  ( $\Delta\epsilon$ ) 266 (−8), 288 (+38), 328 (−3) nm; for <sup>1</sup>H, <sup>13</sup>C, and HMBC NMR spectroscopic data, see Table 1, Table 2, and S1 respectively; HRESIMS  $m/z$  377.1942 [M+H]<sup>+</sup> (calcd. for C<sub>21</sub>H<sub>29</sub>O<sub>6</sub>, 377.1959).

#### 4.8. Talarolutin D (4)

Colorless oil;  $[\alpha]_{\text{D}25} = +214$  ( $c = 0.24$ , CH<sub>3</sub>OH); UV/Vis (CH<sub>3</sub>OH)  $\lambda_{\text{max}}$  (log  $\epsilon$ ) 227 (3.6), 272 (3.7) nm; IR (diamond)  $\nu_{\text{max}}$  1689, 1586, 1421, 1186 cm<sup>-1</sup>; ECD (185  $\mu\text{M}$ , CH<sub>3</sub>OH)  $\lambda_{\text{max}}$  ( $\Delta\epsilon$ ) 238 (+23), 288 (+53), 331 (−12) nm; for <sup>1</sup>H, <sup>13</sup>C, and HMBC NMR spectroscopic data, see Table 1, Table 2, and S1 respectively; HRESIMS  $m/z$  361.1997 [M+H]<sup>+</sup> (calcd. for C<sub>21</sub>H<sub>29</sub>O<sub>5</sub>, 361.2010).

#### 4.9. X-ray diffraction analysis of talarolutin A (1)

A specimen of C<sub>21</sub>H<sub>30</sub>O<sub>5</sub>, approximate dimensions 0.005 mm × 0.010 mm × 0.090 mm, was used for the X-ray crystallographic analysis. The X-ray intensity data were measured. Intensity data were collected at 150K on a D8 goniostat equipped with a Bruker PHOTON100 CMOS detector at Beamline 11.3.1 at the Advanced Light Source (Lawrence Berkeley National Laboratory) using synchrotron radiation tuned to  $\lambda = 1.2399$  Å. For data collection, frames were measured for a duration of 1-s at 0.5° intervals of  $\omega$  with a maximum  $2\theta$  value of ~60°. The total exposure time was 0.73 h. The frames were integrated with the Bruker SAINT software package using a narrow-frame algorithm. The integration of the data using a monoclinic unit cell yielded a total of 11510 reflections to a maximum  $\theta$  angle of 50.93° (0.80 Å resolution), of which 3748 were independent (average redundancy 3.071, completeness = 99.1%,  $R_{\text{int}} = 4.93\%$ ,  $R_{\text{sig}} = 5.08\%$ ) and 3231 (86.21%) were greater than  $2\sigma(F^2)$ . The final cell constants of  $a = 13.0307(13)$  Å,  $b = 5.9960(6)$  Å,  $c = 13.6512(13)$  Å,  $\beta = 118.418(5)^\circ$ , volume = 938.07(16) Å<sup>3</sup>, are based upon the refinement of the XYZ-centroids of 7139 reflections above  $20\sigma(I)$  with  $5.920^\circ < 2\theta < 96.37^\circ$ . Data were corrected for absorption effects using the multi-scan method (SADABS). The ratio of minimum to maximum apparent transmission was 0.804. The calculated minimum and maximum transmission coefficients (based on crystal size) are 0.9670 and 0.9980. The final anisotropic full-matrix least-squares refinement on  $F^2$  with 244 variables converged at  $R1 = 4.15\%$ , for the observed data and  $wR2 = 10.03\%$  for all data. The goodness-of-fit was 1.035. The largest peak in the final difference electron density synthesis was 0.204 e<sup>−</sup>/Å<sup>3</sup> and the largest hole was −0.299 e<sup>−</sup>/Å<sup>3</sup> with an RMS deviation of 0.043 e<sup>−</sup>/Å<sup>3</sup>. On the basis of the final model, the calculated density was 1.280 g/cm<sup>3</sup> and  $F(0\ 0\ 0)$ , 390 e<sup>−</sup>. Crystal data have been deposited with the Cambridge Crystallographic Data Centre and can be accessed using the deposition number, CCDC 1046537.

#### 4.10. Bioassays

Human prostate carcinoma PC-3 cells were obtained from American Type Culture Collection (Manassas, VA) and cultured under ATCC recommended conditions. The effect of pure compounds on viability of PC-3 cells was determined by the methods described previously (Raja et al., 2015). Minimal inhibitory concentrations (MICs) of compounds **2–4** in antimicrobial assays were determined using the methods outlined previously (Ayers et al., 2012). For the quinone reductase assay, the commercially available NQO activity assay kit from Abcam (Cambridge, MA) was utilized exactly as recommended by the vendor.

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#### Supplementary Data

Supplementary data may be found at the end of this formatted document.

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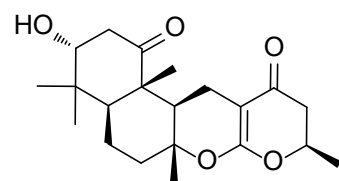
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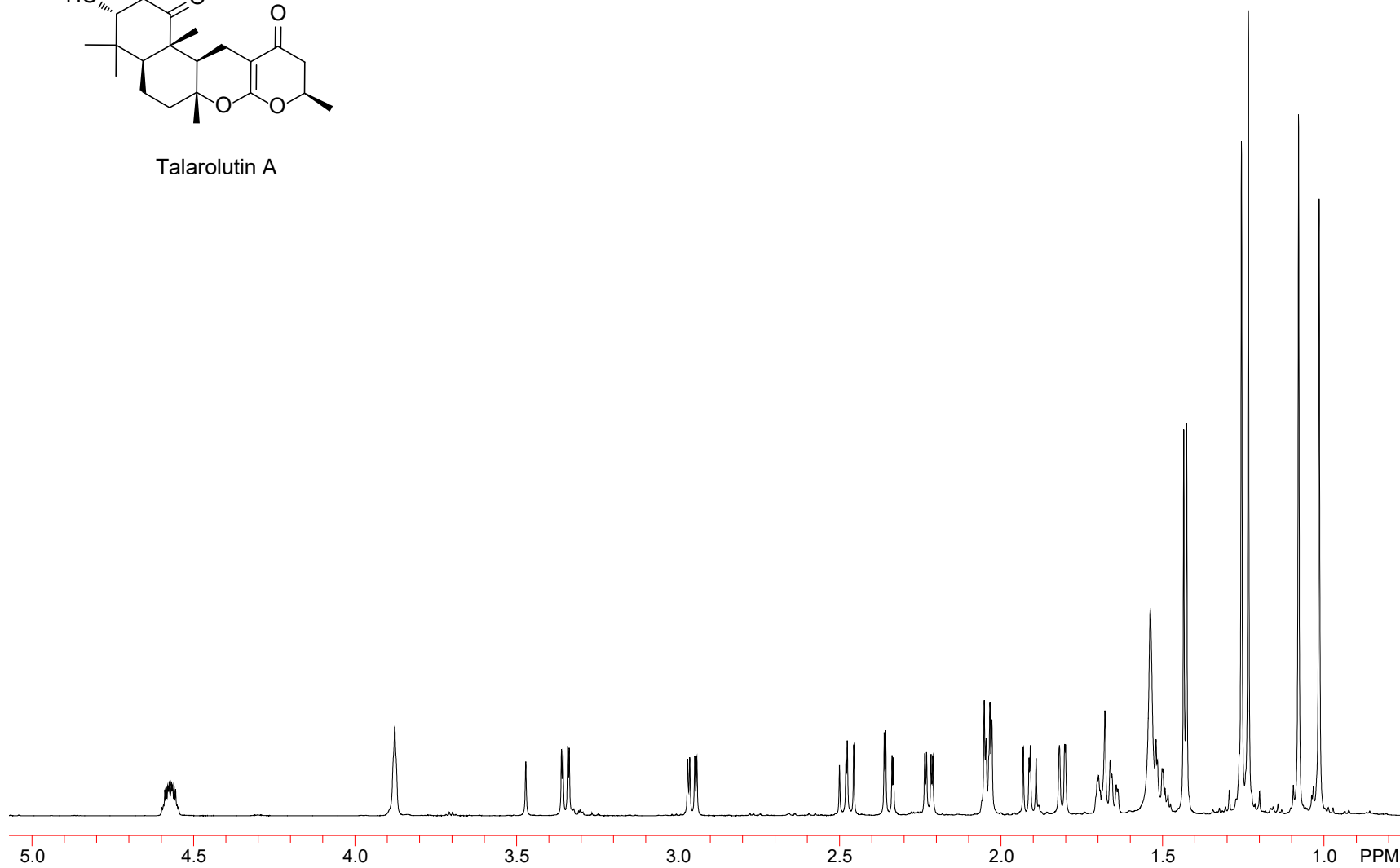
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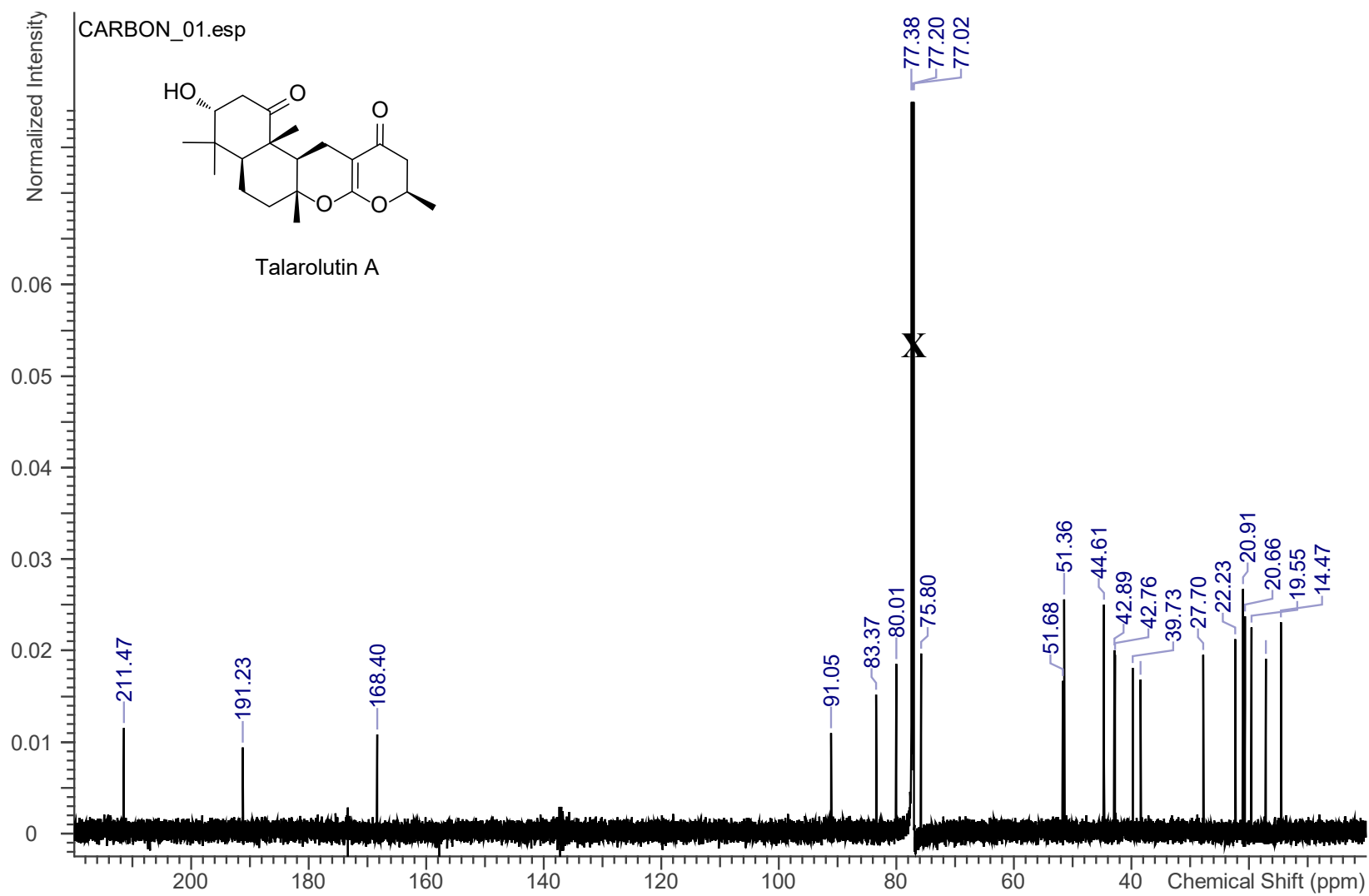
Appendix A. Supplementary data



Talarolutin A



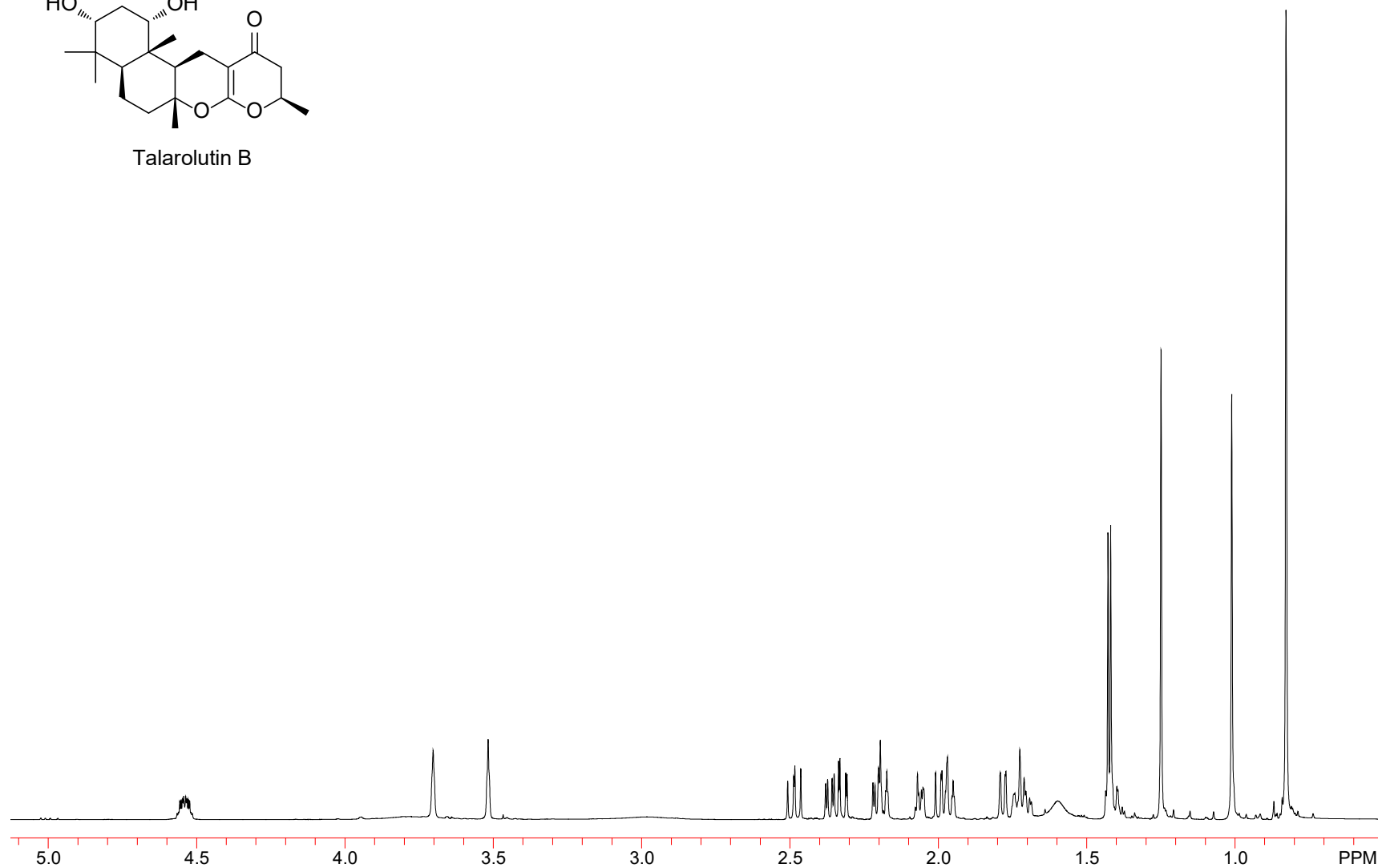
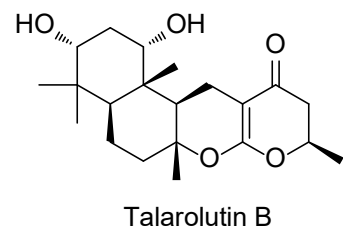
**Figure S1.** <sup>1</sup>H NMR spectrum of talarolutin A (**1**) in CDCl<sub>3</sub> (700 MHz).



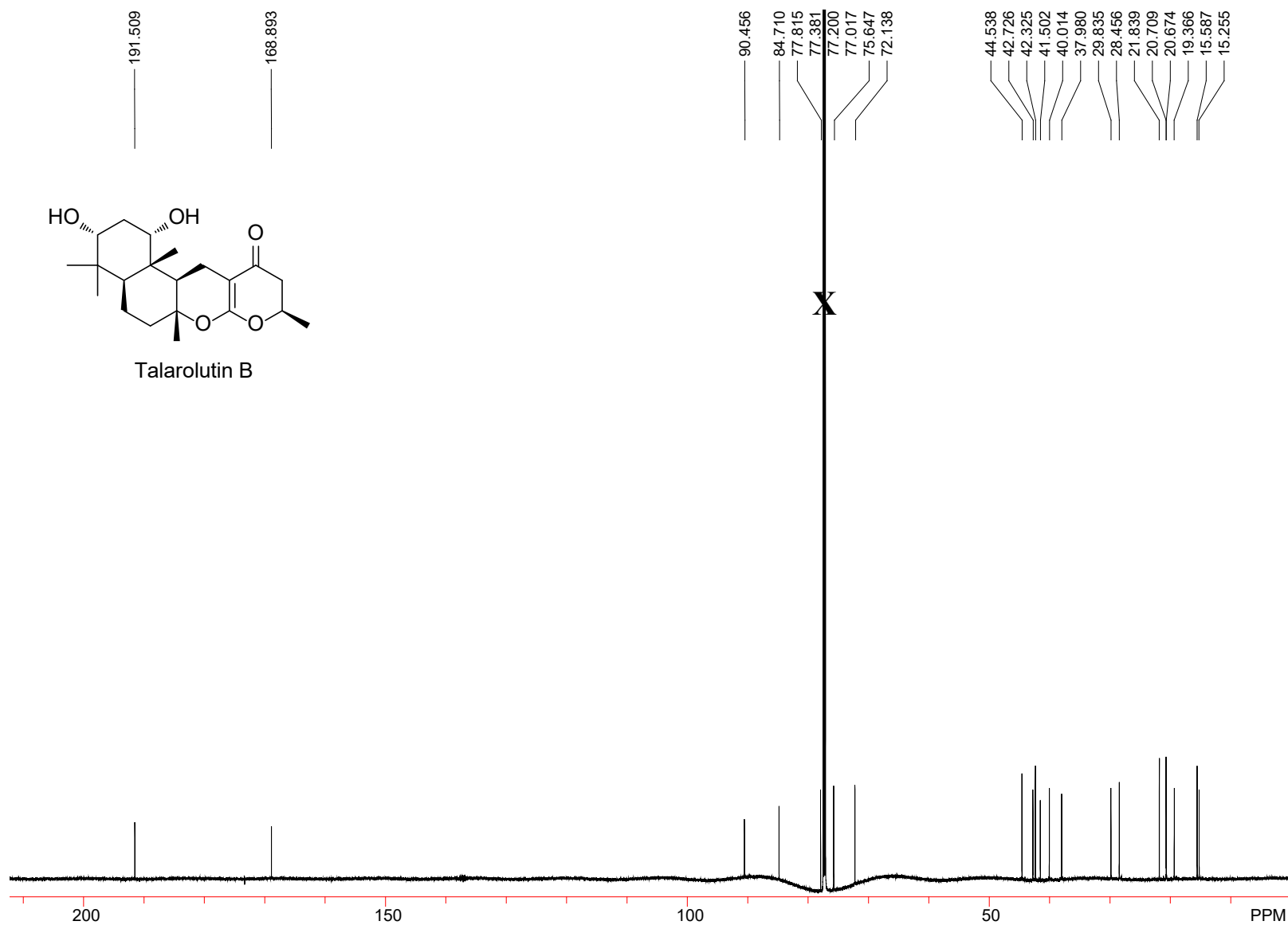
**Figure S2.** <sup>13</sup>C NMR spectrum of talarolutin A (**1**) in CDCl<sub>3</sub> (175 MHz).



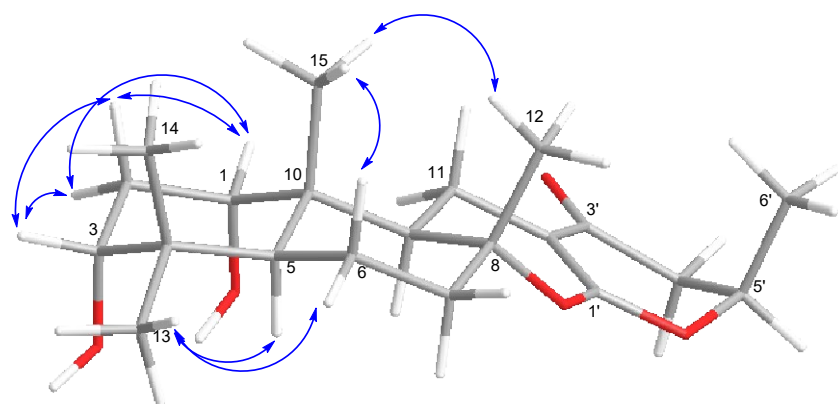




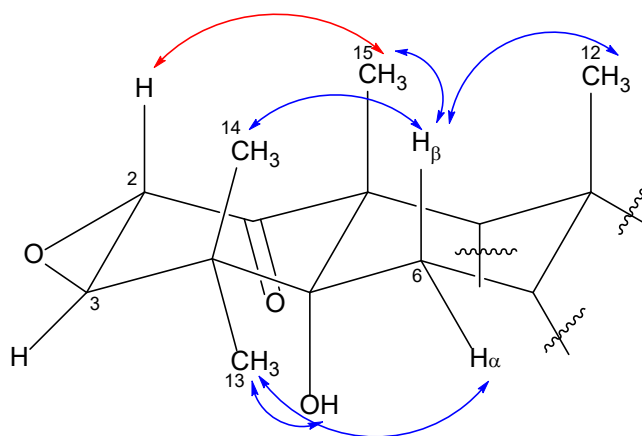
**Figure S4.** <sup>1</sup>H NMR spectrum of talarolutin B (**2**) in CDCl<sub>3</sub> (700 MHz).



**Figure S5.**  $^{13}\text{C}$  NMR spectrum of talarolutin B (**2**) in  $\text{CDCl}_3$  (175 MHz).

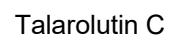


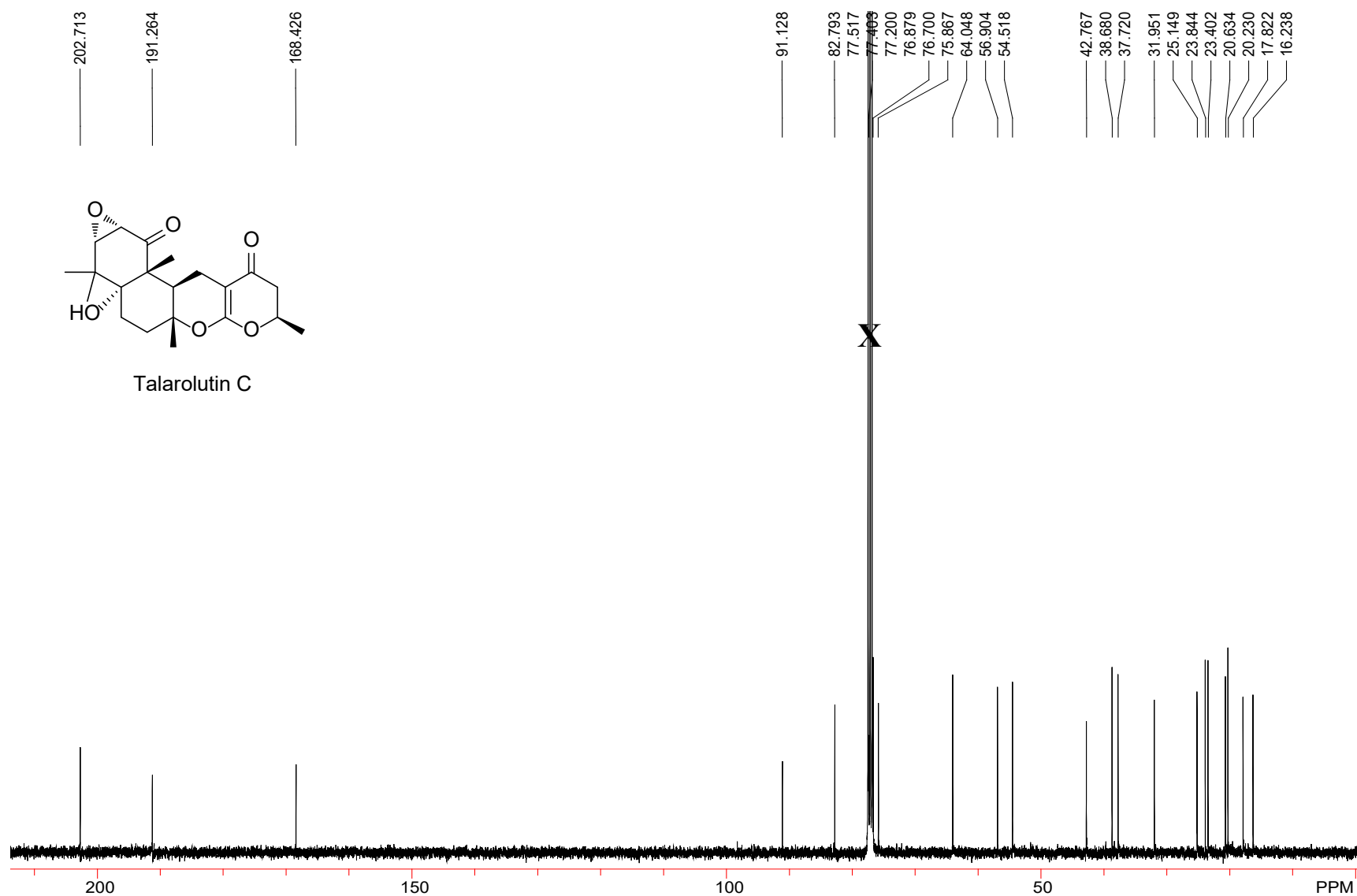
(a)



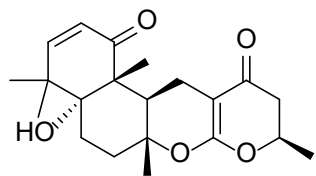
(b)

**Figure S6.** Key NOESY correlations of (a) talarolutins B (**2**) and (b) C (**3**).





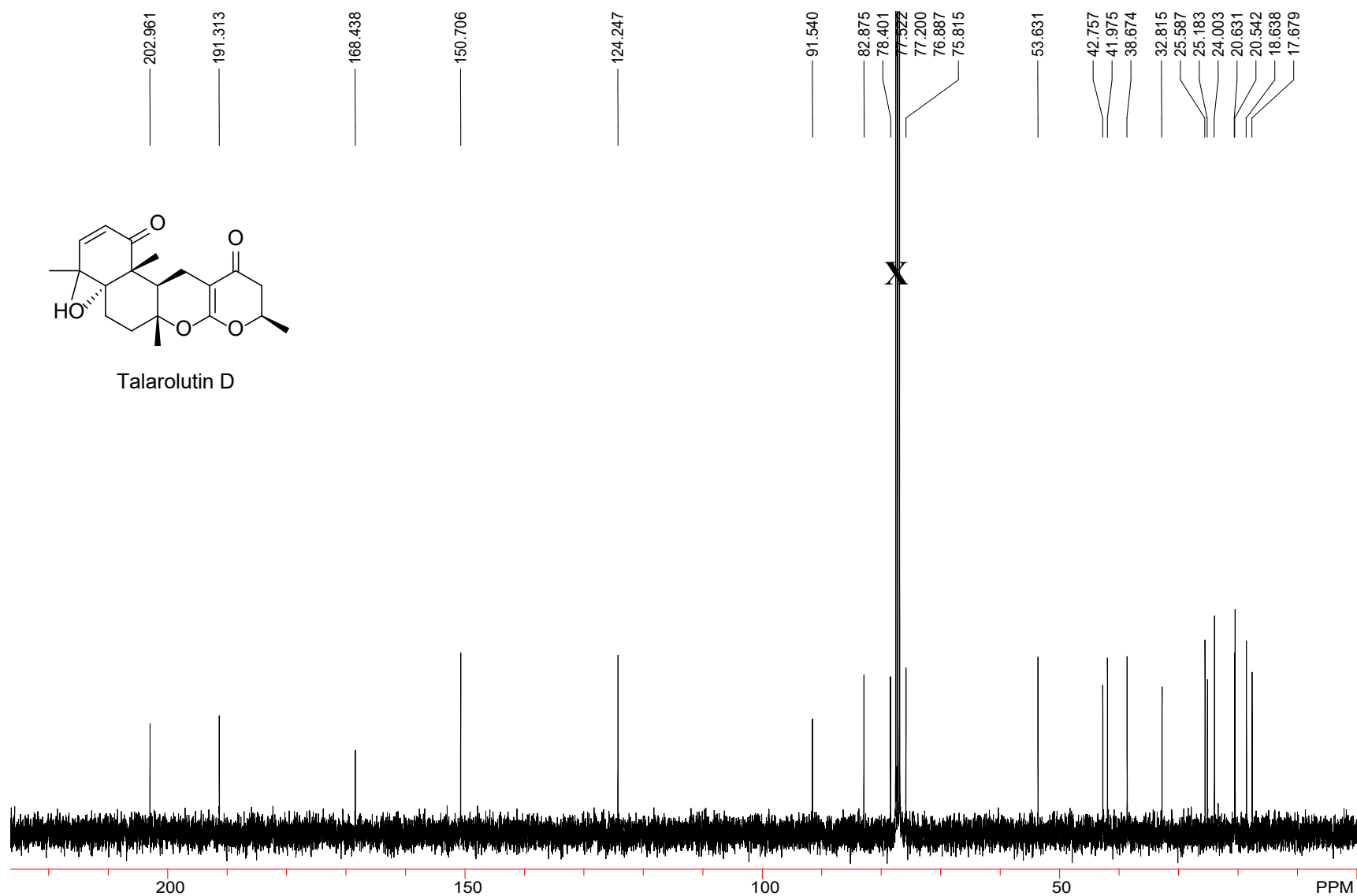
**Figure S8.**  $^{13}\text{C}$  NMR spectrum of talarolutin C (**3**) in  $\text{CDCl}_3$  (100 MHz).



Talarolutin D

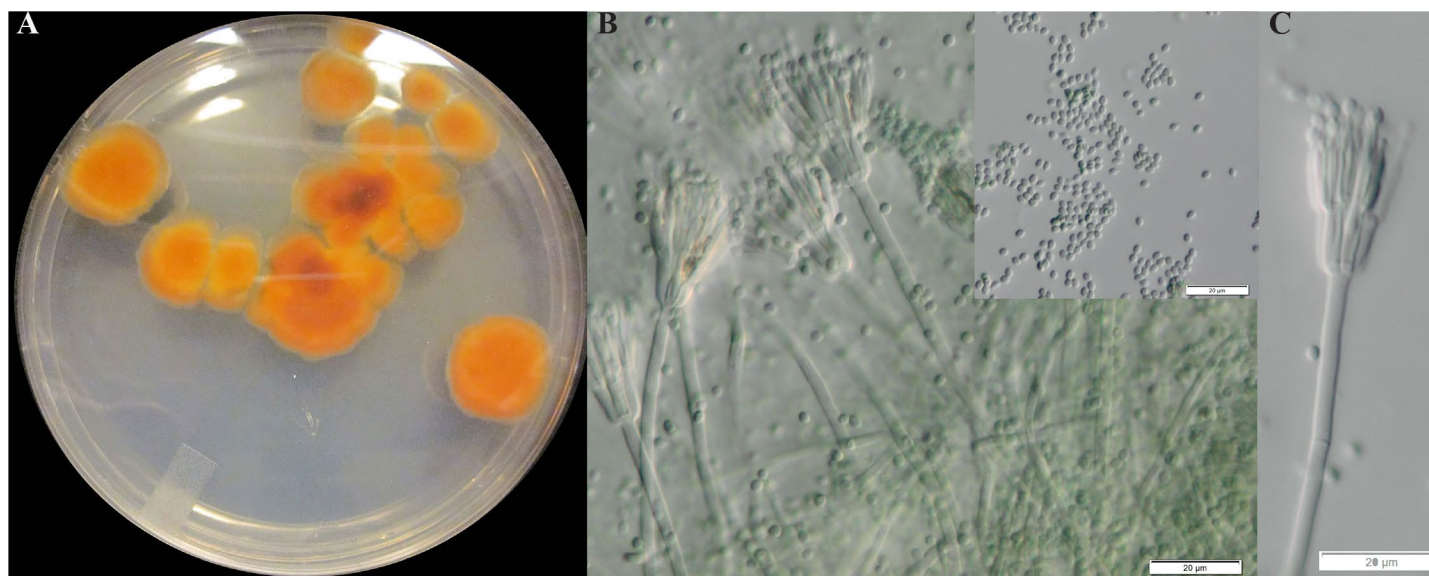


**Figure S9.** <sup>1</sup>H NMR spectrum of talarolutin D (**4**) in CDCl<sub>3</sub> (700 MHz).

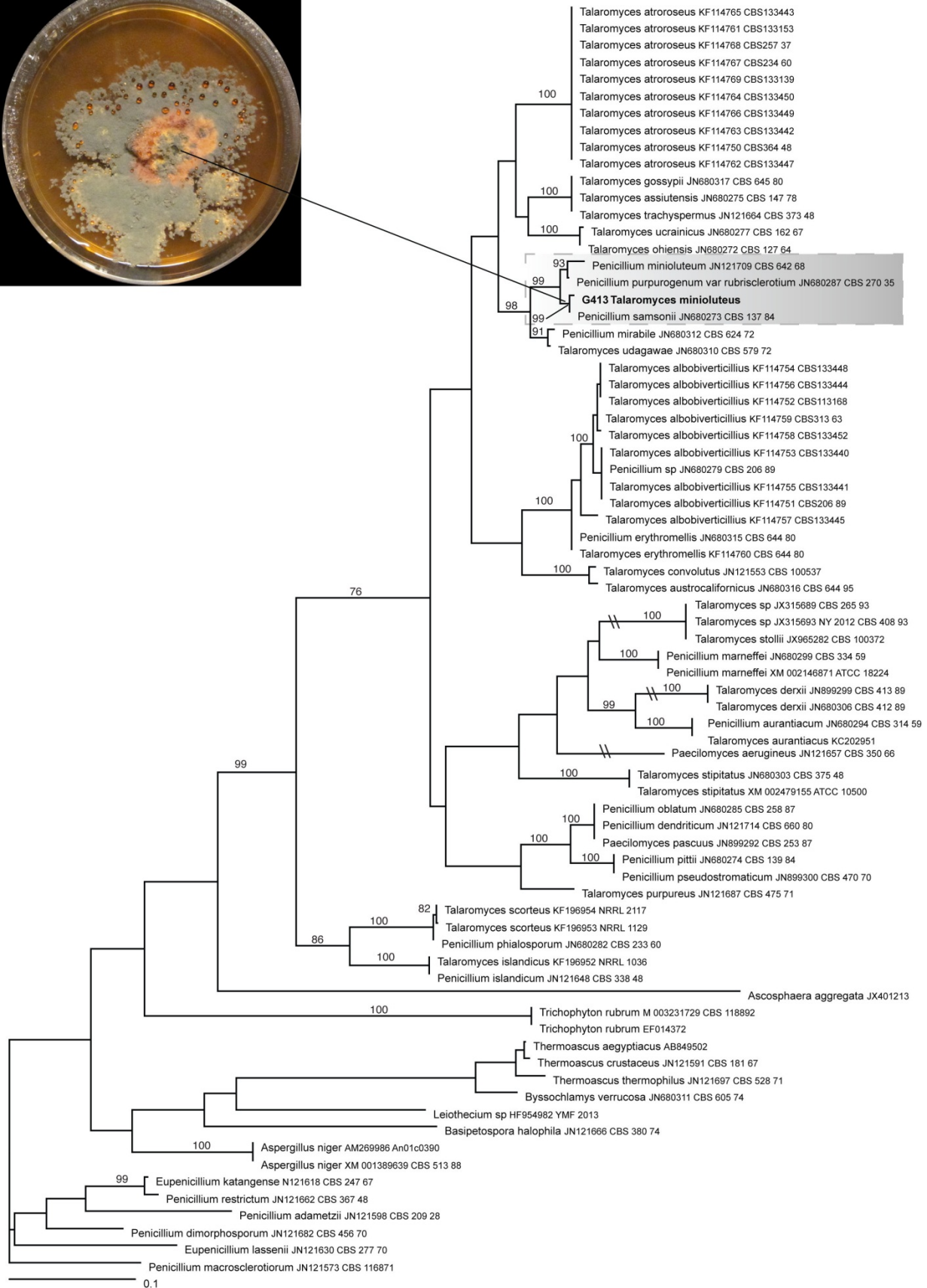
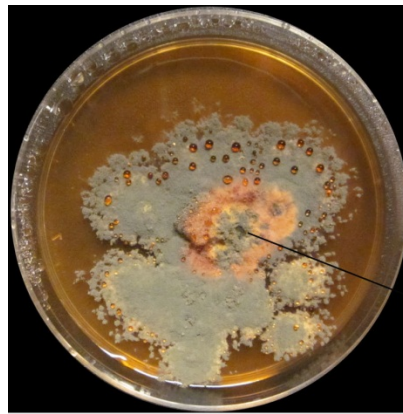


**Figure S10.**  $^{13}\text{C}$  NMR spectrum of talarolutin D (**4**) in  $\text{CDCl}_3$  (100 MHz).





**Figure S11.** **A:** Reverse colony on MEA; brownish orange; **B** and **C:** Biverticillate conidiophores and smooth ellipsoidal conidia. Measure bars = 20  $\mu\text{m}$ .



**Figure S12.** Phylogram of the most likely tree ( $-\ln L = 1561.75$ ) from a RAxML analysis of 75 taxa based on *RPB1* sequence data (785 bp). Numbers refer to RAxML bootstrap support values  $\geq 70\%$  based on 1000 replicates. Strain G413 is identified as *Talaromyces minioluteus* (Bold, highlighted in gray). The symbol (\\) indicates, the branches were shortened for illustration purpose. A 30d old culture on Potato Dextrose Agar media is shown. Bar indicates nucleotide substitution per site.

**Table S1.** HMBC NMR data (400 MHz) of talarolutins A–D (**1–4**) in CDCl<sub>3</sub>.

	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>
#	HMBC (H→ #C)	HMBC (H→ #C)	HMBC (H→ #C)	HMBC (H→ #C)
1		2 (wk), 3, 5, 15 (wk)		
2 $\alpha$	1, 3		1, 10	1 (wk), 4, 10, 13 (wk), 14 (wk)
2 $\beta$	1, 3, 4, 10	1, 3, 4, 10		
3	1, 5, 13, 14 (wk)	1, 2 (wk), 5, 13 (wk), 14 (wk)	1, 4, 5, 14	1, 4, 5, 13, 14
5	1 (wk), 3 (wk), 4, 6, 7, 9, 10, 13, 14, 15	4, 6, 7, 9 (wk), 13, 14 (wk), 15		
6 $\alpha$	4, 5, 8, 10	5 (wk), 8, 10	5, 7, 8, 10 (wk)	5, 8, 9
6 $\beta$	5, 7, 8, 10		7, 8 (wk)	5, 7, 8, 10
7 $\alpha$	8, 9, 12	5, 8 (wk), 9, 12 (wk)	6, 8, 12	6, 8, 11 (wk), 12
7 $\beta$	5, 6, 8, 12	6, 8, 12	5, 6, 8	5, 9, 11 (wk), 12
9	1, 7 (wk), 8, 10, 11, 12, 15, 2' (wk)	1 (wk), 5 (wk), 8, 10, 12, 15, 2' (wk)	1, 8, 10, 11, 12, 15	1, 8, 10, 12, 15
11 $\alpha$	8, 9, 1', 2', 3'	8, 9, 1', 2'	8, 9, 1', 2', 3'	8, 9, 1', 2', 3'
11 $\beta$	8 (wk), 9, 10, 1', 2'	9, 10, 1', 2'	9, 10 (wk), 1', 2'	8, 9, 10, 1', 2'
12	7, 8, 9	7, 8, 9	7, 8, 9	7, 8, 9
13	3, 4, 5, 14	3, 4, 5, 14	3, 4, 5, 14	3, 4, 5, 14
14	3, 4, 5, 13	3, 4, 5, 13	3, 4, 5, 13	3, 4, 5, 13
15	1, 5, 9, 10	1, 5, 9, 10	1, 5, 9, 10	1, 5, 9, 10
4' $\alpha$	5', 3', 6'	5', 3', 6'	5', 3', 6'	5', 3', 6'
4' $\beta$	2' (wk), 3'	2', 3'	2', 3'	2' (wk), 3'
5'	3' (wk), 6' (wk)	3' (wk), 6' (wk)	3' (wk), 6' (wk)	
6'	1' (wk), 3' (wk), 4', 5'	1' (wk), 3' (wk), 4', 5'	1' (wk), 3' (wk), 4', 5'	1' (wk), 3' (wk), 4', 5'
5-OH			4, 5, 6	

## X-ray Crystallographic Analysis of 1

**Table S2.** Sample and crystal data for **1**.

Identification code	Talarolutin A	
Chemical formula	$C_{21}H_{30}O_5$	
Formula weight	362.45	
Temperature	150(2) K	
Wavelength	1.23990 Å	
Crystal size	0.005 x 0.010 x 0.090 mm	
Crystal system	monoclinic	
Space group	P 1 21 1	
Unit cell dimensions	a = 13.0307(13) Å	$\alpha = 90^\circ$
	b = 5.9960(6) Å	$\beta = 118.418(5)^\circ$
	c = 13.6512(13) Å	$\gamma = 90^\circ$
Volume	938.07(16) Å <sup>3</sup>	
Z	2	
Density (calculated)	1.283 g/cm <sup>3</sup>	
Absorption coefficient	0.371 mm <sup>-1</sup>	
F(000)	392	

**Table S3.** Data collection and structure refinement for **1**.

Theta range for data collection	2.96 to 50.93°	
Index ranges	-16<= <i>h</i> <=16, -7<= <i>k</i> <=7, -17<= <i>l</i> <=17	
Reflections collected	11510	
Independent reflections	3748 [R(int) = 0.0493]	
Coverage of independent reflections	99.1%	
Absorption correction	multi-scan	
Max. and min. transmission	0.9980 and 0.9670	
Refinement method	Full-matrix least-squares on F <sup>2</sup>	
Refinement program	SHELXL-2013 (Sheldrick, 2013)	
Function minimized	$\Sigma w(F_o^2 - F_c^2)^2$	
Data / restraints / parameters	3748 / 1 / 244	
Goodness-of-fit on F <sup>2</sup>	1.035	
$\Delta/\sigma_{\max}$	0.001	
Final R indices	3231 data; I>2 $\sigma$ (I)	R1 = 0.0415, wR2 = 0.0960
	all data	R1 = 0.0513, wR2 = 0.1003
Weighting scheme	w=1/[ $\sigma^2(F_o^2)+(0.0540P)^2+0.0527P$ ] where P=(F <sub>o</sub> <sup>2</sup> +2F <sub>c</sub> <sup>2</sup> )/3	
Absolute structure parameter	0.01(18)	
Largest diff. peak and hole	0.204 and -0.299 eÅ <sup>-3</sup>	
R.M.S. deviation from mean	0.043 eÅ <sup>-3</sup>	

**Table S4.** Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic atomic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **1**.

U(eq) is defined as one third of the trace of the orthogonalized  $U_{ij}$  tensor.

	x/a	y/b	z/c	U(eq)
C(1)	3852(2)	1100(4)	9011(2)	24(1)
O(1)	4603(2)	2347(3)	9644(1)	31(1)
C(2)	2875(2)	324(5)	9235(2)	30(1)
C(3)	1730(2)	1411(5)	8375(2)	32(1)
O(3)	1805(2)	3775(4)	8507(2)	39(1)
C(4)	1443(2)	904(5)	7163(2)	28(1)
C(5)	2533(2)	1536(4)	7018(2)	24(1)
C(6)	2343(2)	1334(5)	5821(2)	29(1)
C(7)	3314(2)	2594(5)	5716(2)	29(1)
C(8)	4515(2)	1699(4)	6519(2)	28(1)
O(8)	5321(1)	3483(3)	6525(1)	31(1)
C(9)	4712(2)	1590(4)	7724(2)	24(1)
C(10)	3717(2)	439(4)	7864(2)	23(1)
C(11)	5949(2)	740(4)	8483(2)	26(1)
C(1')	6412(2)	3421(5)	7358(2)	30(1)
C(2')	6802(2)	2086(5)	8270(2)	28(1)
C(3')	8036(2)	1871(5)	8973(2)	31(1)
O(3')	8483(2)	513(4)	9750(2)	41(1)
C(4')	8801(2)	3344(5)	8685(2)	38(1)
C(5')	8173(2)	5450(5)	8115(2)	36(1)
O(5')	7057(2)	4883(3)	7140(2)	34(1)
C(12)	4805(3)	-387(5)	6068(2)	36(1)
C(13)	396(2)	2354(6)	6377(2)	39(1)
C(14)	1053(2)	-1537(5)	6887(2)	36(1)
C(15)	3746(2)	-2131(4)	7768(2)	28(1)
C(6')	8821(3)	6900(6)	7689(3)	46(1)

**Table S5.** Bond lengths (Å) for **1**.

C(1)-O(1)	1.208(3)	C(11)-C(2')	1.509(3)
C(1)-C(2)	1.517(3)	C(11)-H(11A)	0.99
C(1)-C(10)	1.542(3)	C(11)-H(11B)	0.99
C(2)-C(3)	1.536(4)	C(1')-O(5')	1.342(3)
C(2)-H(2A)	0.99	C(1')-C(2')	1.359(4)
C(2)-H(2B)	0.99	C(2')-C(3')	1.434(4)
C(3)-O(3)	1.426(4)	C(3')-O(3')	1.241(3)
C(3)-C(4)	1.543(4)	C(3')-C(4')	1.516(4)
C(3)-H(3)	1	C(4')-C(5')	1.505(4)
O(3)-H(3A)	0.88(3)	C(4')-H(4'A)	0.99
C(4)-C(14)	1.537(4)	C(4')-H(4'B)	0.99
C(4)-C(13)	1.542(4)	C(5')-O(5')	1.469(3)
C(4)-C(5)	1.571(3)	C(5')-C(6')	1.508(4)
C(5)-C(6)	1.537(3)	C(5')-H(5')	1
C(5)-C(10)	1.566(3)	C(12)-H(12A)	0.98
C(5)-H(5)	1	C(12)-H(12B)	0.98
C(6)-C(7)	1.537(3)	C(12)-H(12C)	0.98
C(6)-H(6A)	0.99	C(13)-H(13A)	0.98
C(6)-H(6B)	0.99	C(13)-H(13B)	0.98
C(7)-C(8)	1.519(4)	C(13)-H(13C)	0.98
C(7)-H(7A)	0.99	C(14)-H(14A)	0.98
C(7)-H(7B)	0.99	C(14)-H(14B)	0.98
C(8)-O(8)	1.497(3)	C(14)-H(14C)	0.98
C(8)-C(12)	1.519(4)	C(15)-H(15A)	0.98
C(8)-C(9)	1.541(3)	C(15)-H(15B)	0.98
O(8)-C(1')	1.333(3)	C(15)-H(15C)	0.98
C(9)-C(11)	1.529(4)	C(6')-H(6'A)	0.98
C(9)-C(10)	1.559(3)	C(6')-H(6'B)	0.98
C(9)-H(9)	1	C(6')-H(6'C)	0.98
C(10)-C(15)	1.549(4)		

**Table S6.** Bond angles (°) for **1**.

O(1)-C(1)-C(2)	121.2(2)	C(2')-C(11)-C(9)	109.1(2)
O(1)-C(1)-C(10)	123.2(2)	C(2')-C(11)-H(11A)	109.9
C(2)-C(1)-C(10)	115.1(2)	C(9)-C(11)-H(11A)	109.9
C(1)-C(2)-C(3)	108.6(2)	C(2')-C(11)-H(11B)	109.9
C(1)-C(2)-H(2A)	110	C(9)-C(11)-H(11B)	109.9
C(3)-C(2)-H(2A)	110	H(11A)-C(11)-H(11B)	108.3
C(1)-C(2)-H(2B)	110	O(8)-C(1')-O(5')	108.6(2)
C(3)-C(2)-H(2B)	110	O(8)-C(1')-C(2')	125.5(2)
H(2A)-C(2)-H(2B)	108.3	O(5')-C(1')-C(2')	125.9(2)
O(3)-C(3)-C(2)	110.0(2)	C(1')-C(2')-C(3')	118.5(2)
O(3)-C(3)-C(4)	107.3(2)	C(1')-C(2')-C(11)	120.2(2)
C(2)-C(3)-C(4)	112.8(2)	C(3')-C(2')-C(11)	121.2(2)
O(3)-C(3)-H(3)	108.9	O(3')-C(3')-C(2')	123.5(2)

C(2)-C(3)-H(3)	108.9	O(3')-C(3')-C(4')	120.3(2)
C(4)-C(3)-H(3)	108.9	C(2')-C(3')-C(4')	116.2(2)
C(3)-O(3)-H(3A)	109(2)	C(5')-C(4')-C(3')	111.1(2)
C(14)-C(4)-C(13)	106.6(2)	C(5')-C(4')-H(4'A)	109.4
C(14)-C(4)-C(3)	109.5(2)	C(3')-C(4')-H(4'A)	109.4
C(13)-C(4)-C(3)	108.5(2)	C(5')-C(4')-H(4'B)	109.4
C(14)-C(4)-C(5)	114.3(2)	C(3')-C(4')-H(4'B)	109.4
C(13)-C(4)-C(5)	109.3(2)	H(4'A)-C(4')-H(4'B)	108
C(3)-C(4)-C(5)	108.4(2)	O(5')-C(5')-C(4')	109.5(2)
C(6)-C(5)-C(10)	110.8(2)	O(5')-C(5')-C(6')	106.2(2)
C(6)-C(5)-C(4)	114.7(2)	C(4')-C(5')-C(6')	114.7(2)
C(10)-C(5)-C(4)	116.0(2)	O(5')-C(5')-H(5')	108.8
C(6)-C(5)-H(5)	104.7	C(4')-C(5')-H(5')	108.8
C(10)-C(5)-H(5)	104.7	C(6')-C(5')-H(5')	108.8
C(4)-C(5)-H(5)	104.7	C(1')-O(5')-C(5')	113.89(19)
C(5)-C(6)-C(7)	109.7(2)	C(8)-C(12)-H(12A)	109.5
C(5)-C(6)-H(6A)	109.7	C(8)-C(12)-H(12B)	109.5
C(7)-C(6)-H(6A)	109.7	H(12A)-C(12)-H(12B)	109.5
C(5)-C(6)-H(6B)	109.7	C(8)-C(12)-H(12C)	109.5
C(7)-C(6)-H(6B)	109.7	H(12A)-C(12)-H(12C)	109.5
H(6A)-C(6)-H(6B)	108.2	H(12B)-C(12)-H(12C)	109.5
C(8)-C(7)-C(6)	111.7(2)	C(4)-C(13)-H(13A)	109.5
C(8)-C(7)-H(7A)	109.3	C(4)-C(13)-H(13B)	109.5
C(6)-C(7)-H(7A)	109.3	H(13A)-C(13)-H(13B)	109.5
C(8)-C(7)-H(7B)	109.3	C(4)-C(13)-H(13C)	109.5
C(6)-C(7)-H(7B)	109.3	H(13A)-C(13)-H(13C)	109.5
H(7A)-C(7)-H(7B)	107.9	H(13B)-C(13)-H(13C)	109.5
O(8)-C(8)-C(7)	103.4(2)	C(4)-C(14)-H(14A)	109.5
O(8)-C(8)-C(12)	106.33(19)	C(4)-C(14)-H(14B)	109.5
C(7)-C(8)-C(12)	111.6(2)	H(14A)-C(14)-H(14B)	109.5
O(8)-C(8)-C(9)	105.37(19)	C(4)-C(14)-H(14C)	109.5
C(7)-C(8)-C(9)	112.1(2)	H(14A)-C(14)-H(14C)	109.5
C(12)-C(8)-C(9)	116.8(2)	H(14B)-C(14)-H(14C)	109.5
C(1')-O(8)-C(8)	116.4(2)	C(10)-C(15)-H(15A)	109.5
C(11)-C(9)-C(8)	108.90(19)	C(10)-C(15)-H(15B)	109.5
C(11)-C(9)-C(10)	116.1(2)	H(15A)-C(15)-H(15B)	109.5
C(8)-C(9)-C(10)	115.1(2)	C(10)-C(15)-H(15C)	109.5
C(11)-C(9)-H(9)	105.2	H(15A)-C(15)-H(15C)	109.5
C(8)-C(9)-H(9)	105.2	H(15B)-C(15)-H(15C)	109.5
C(10)-C(9)-H(9)	105.2	C(5')-C(6')-H(6'A)	109.5
C(1)-C(10)-C(15)	110.2(2)	C(5')-C(6')-H(6'B)	109.5
C(1)-C(10)-C(9)	108.2(2)	H(6'A)-C(6')-H(6'B)	109.5
C(15)-C(10)-C(9)	112.0(2)	C(5')-C(6')-H(6'C)	109.5
C(1)-C(10)-C(5)	103.72(19)	H(6'A)-C(6')-H(6'C)	109.5
C(15)-C(10)-C(5)	114.2(2)	H(6'B)-C(6')-H(6'C)	109.5
C(9)-C(10)-C(5)	108.0(2)		

**Table S7.** Torsion angles (°) for **1**.

O(1)-C(1)-C(2)-C(3)	110.2(3)	C(2)-C(1)-C(10)-C(5)	59.2(3)
C(10)-C(1)-C(2)-C(3)	-62.1(3)	C(11)-C(9)-C(10)-C(1)	68.5(3)
C(1)-C(2)-C(3)-O(3)	-63.2(3)	C(8)-C(9)-C(10)-C(1)	-162.7(2)
C(1)-C(2)-C(3)-C(4)	56.6(3)	C(11)-C(9)-C(10)-C(15)	-53.2(3)
O(3)-C(3)-C(4)-C(14)	-166.0(2)	C(8)-C(9)-C(10)-C(15)	75.6(3)
C(2)-C(3)-C(4)-C(14)	72.7(3)	C(11)-C(9)-C(10)-C(5)	-179.9(2)
O(3)-C(3)-C(4)-C(13)	-50.0(3)	C(8)-C(9)-C(10)-C(5)	-51.0(3)
C(2)-C(3)-C(4)-C(13)	-171.2(2)	C(6)-C(5)-C(10)-C(1)	171.7(2)
O(3)-C(3)-C(4)-C(5)	68.6(3)	C(4)-C(5)-C(10)-C(1)	-55.3(3)
C(2)-C(3)-C(4)-C(5)	-52.7(3)	C(6)-C(5)-C(10)-C(15)	-68.2(3)
C(14)-C(4)-C(5)-C(6)	63.0(3)	C(4)-C(5)-C(10)-C(15)	64.8(3)
C(13)-C(4)-C(5)-C(6)	-56.4(3)	C(6)-C(5)-C(10)-C(9)	57.1(3)
C(3)-C(4)-C(5)-C(6)	-174.5(2)	C(4)-C(5)-C(10)-C(9)	-169.89(19)
C(14)-C(4)-C(5)-C(10)	-68.2(3)	C(8)-C(9)-C(11)-C(2')	50.4(3)
C(13)-C(4)-C(5)-C(10)	172.4(2)	C(10)-C(9)-C(11)-C(2')	-177.8(2)
C(3)-C(4)-C(5)-C(10)	54.3(3)	C(8)-O(8)-C(1')-O(5')	170.0(2)
C(10)-C(5)-C(6)-C(7)	-62.2(3)	C(8)-O(8)-C(1')-C(2')	-9.6(4)
C(4)-C(5)-C(6)-C(7)	164.1(2)	O(8)-C(1')-C(2')-C(3')	167.1(2)
C(5)-C(6)-C(7)-C(8)	58.8(3)	O(5')-C(1')-C(2')-C(3')	-12.4(4)
C(6)-C(7)-C(8)-O(8)	-164.7(2)	O(8)-C(1')-C(2')-C(11)	-8.5(4)
C(6)-C(7)-C(8)-C(12)	81.4(3)	O(5')-C(1')-C(2')-C(11)	171.9(2)
C(6)-C(7)-C(8)-C(9)	-51.7(3)	C(9)-C(11)-C(2')-C(1')	-13.4(3)
C(7)-C(8)-O(8)-C(1')	163.9(2)	C(9)-C(11)-C(2')-C(3')	171.0(2)
C(12)-C(8)-O(8)-C(1')	-78.5(3)	C(1')-C(2')-C(3')-O(3')	-171.7(3)
C(9)-C(8)-O(8)-C(1')	46.1(3)	C(11)-C(2')-C(3')-O(3')	3.9(4)
O(8)-C(8)-C(9)-C(11)	-66.6(2)	C(1')-C(2')-C(3')-C(4')	5.5(4)
C(7)-C(8)-C(9)-C(11)	-178.4(2)	C(11)-C(2')-C(3')-C(4')	-178.9(2)
C(12)-C(8)-C(9)-C(11)	51.1(3)	O(3')-C(3')-C(4')-C(5')	-155.0(3)
O(8)-C(8)-C(9)-C(10)	161.1(2)	C(2')-C(3')-C(4')-C(5')	27.8(3)
C(7)-C(8)-C(9)-C(10)	49.3(3)	C(3')-C(4')-C(5')-O(5')	-54.6(3)
C(12)-C(8)-C(9)-C(10)	-81.2(3)	C(3')-C(4')-C(5')-C(6')	-173.7(3)
O(1)-C(1)-C(10)-C(15)	124.3(3)	O(8)-C(1')-O(5')-C(5')	163.5(2)
C(2)-C(1)-C(10)-C(15)	-63.5(3)	C(2')-C(1')-O(5')-C(5')	-16.9(4)
O(1)-C(1)-C(10)-C(9)	1.6(3)	C(4')-C(5')-O(5')-C(1')	50.0(3)
C(2)-C(1)-C(10)-C(9)	173.7(2)	C(6')-C(5')-O(5')-C(1')	174.3(2)
O(1)-C(1)-C(10)-C(5)	-112.9(3)		



**Table S8.** Anisotropic atomic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **1**.

The anisotropic atomic displacement factor exponent takes the form:  $-2\pi^2 [h^2 a^{*2} U_{11} + \dots + 2 h k a^* b^* U_{12}]$

	U <sub>11</sub>	U <sub>22</sub>	U <sub>33</sub>	U <sub>23</sub>	U <sub>13</sub>	U <sub>12</sub>
C(1)	26(1)	20(1)	24(1)	3(1)	10(1)	3(1)
O(1)	32(1)	33(1)	28(1)	-7(1)	13(1)	-6(1)
C(2)	33(1)	34(2)	25(1)	0(1)	16(1)	-2(1)
C(3)	30(2)	35(2)	38(2)	2(1)	22(1)	1(1)
O(3)	50(1)	36(1)	38(1)	1(1)	27(1)	7(1)
C(4)	24(1)	33(2)	30(1)	3(1)	14(1)	-3(1)
C(5)	24(1)	24(1)	25(1)	1(1)	12(1)	-4(1)
C(6)	26(1)	36(2)	24(1)	1(1)	11(1)	-3(1)
C(7)	29(1)	35(2)	22(1)	2(1)	11(1)	-5(1)
C(8)	28(1)	32(2)	27(1)	2(1)	16(1)	-5(1)
O(8)	25(1)	38(1)	27(1)	6(1)	12(1)	-6(1)
C(9)	25(1)	22(1)	25(1)	-1(1)	12(1)	-3(1)
C(10)	23(1)	24(1)	24(1)	0(1)	12(1)	-3(1)
C(11)	27(1)	29(1)	25(1)	2(1)	14(1)	-2(1)
C(1')	27(1)	36(2)	31(1)	-2(1)	17(1)	-4(1)
C(2')	26(1)	33(2)	28(1)	-1(1)	15(1)	-2(1)
C(3')	29(1)	38(2)	32(1)	0(1)	19(1)	-1(1)
O(3')	28(1)	54(1)	42(1)	15(1)	17(1)	3(1)
C(4')	26(1)	51(2)	38(2)	3(1)	16(1)	-6(1)
C(5')	30(2)	42(2)	36(2)	1(1)	16(1)	-8(1)
O(5')	28(1)	42(1)	34(1)	7(1)	17(1)	-6(1)
C(12)	39(2)	42(2)	30(1)	-3(1)	21(1)	-1(1)
C(13)	28(1)	51(2)	39(2)	7(2)	17(1)	2(1)
C(14)	31(2)	42(2)	35(2)	-3(1)	15(1)	-13(1)
C(15)	31(1)	24(2)	30(1)	-2(1)	16(1)	-4(1)
C(6')	36(2)	53(2)	52(2)	3(2)	22(2)	-13(2)

**Table S9.** Hydrogen atomic coordinates ( $\times 10^4$ ) and isotropic atomic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **1**.

	<b>x</b>	<b>y</b>	<b>z</b>	<b>U(eq)</b>
H(2A)	3044	758	9997	36
H(2B)	2805	-1320	9174	36
H(3)	1078	853	8499	38
H(3A)	1850(30)	4120(60)	9150(30)	53(10)
H(5)	2649	3168	7187	29
H(6A)	1574	1966	5294	35
H(6B)	2352	-258	5632	35
H(7A)	3187	2453	4944	35
H(7B)	3274	4197	5869	35
H(9)	4713	3177	7949	29
H(11A)	6135	891	9273	32
H(11B)	6006	-856	8330	32
H(4'A)	9521	3729	9375	45
H(4'B)	9028	2518	8190	45
H(5')	8012	6339	8647	43
H(12A)	4760	-54	5346	53
H(12B)	4245	-1567	5976	53
H(12C)	5595	-885	6591	53
H(13A)	-219	2249	6595	58
H(13B)	94	1825	5610	58
H(13C)	646	3910	6427	58
H(14A)	247	-1687	6753	54
H(14B)	1565	-2491	7513	54
H(14C)	1097	-1990	6218	54
H(15A)	4558	-2644	8131	41
H(15B)	3366	-2558	6980	41
H(15C)	3333	-2817	8132	41
H(6'A)	8351	8223	7326	69
H(6'B)	8964	6057	7151	69
H(6'C)	9567	7362	8313	69

**Table S10.** Hydrogen bond distances ( $\text{\AA}$ ) and angles ( $^\circ$ ) for **1**.

	<b>Donor-H</b>	<b>Acceptor-H</b>	<b>Donor-Acceptor</b>	<b>Angle</b>
O3-H3A $\cdots$ O3' #1	0.88(3)	1.94(4)	2.782(3)	161 (3)

Symmetry transformations used to generate equivalent atoms:

#1  $-x+1, y+1/2, -z+2$